LYOTROPIC LIQUID CRYSTAL – BUTYL RUBBER BLENDED NANOMATERIALS

<u>Jizhu Jin¹</u>, Vinh Nguyen², Xiaoyun Lu¹, Brian J. Elliott*², and Douglas L. Gin*¹

¹Dept. of Chemistry & Biochemistry, and Dept. of Chemical & Biological Engineering, University of Colorado, Boulder, CO 80309

²TDA Research Inc., 12345 W. 52nd Ave., Wheat Ridge, CO 80033

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Introduction

Uses of Butyl Rubber (BR) as barrier material fabric

Advantages:

- Low permeability toward gases, organic solvents, water, and reactive chemicals
- Excellent chemical resistance

Disadvantages:

• Lack of permeability of air and water vapor

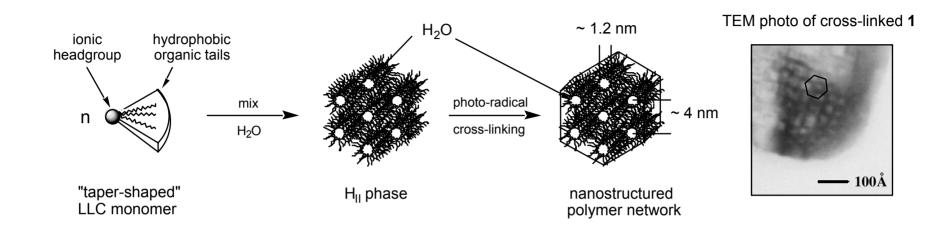
Development of fatigue and heat stress in wearer

Butyl Rubber (BR)

Water Vapor
And Air

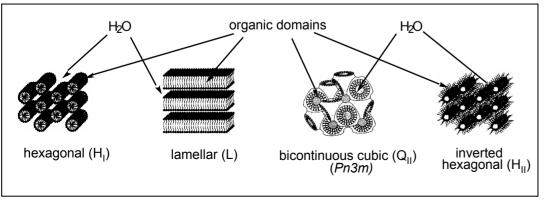
Chemical
Warfare Agent

Our Approach

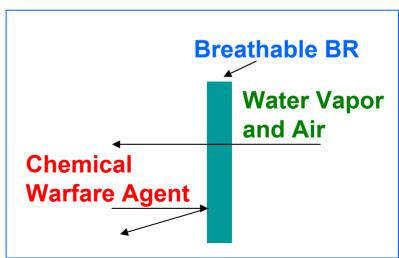


• Blend BR with lyotropic liquid crystals (LLCs) that form inverted hexagonal (H_{II}) phase





LLC assemblies

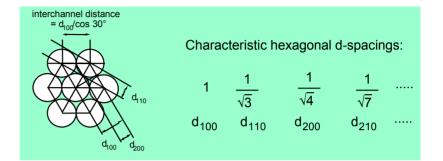


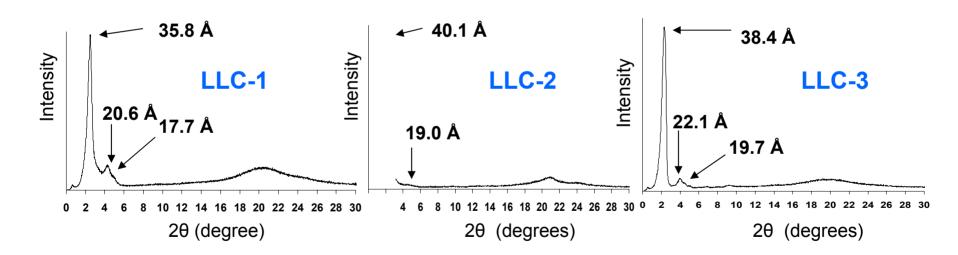
Structure of LLCs and X-ray Diffraction Characterization

OR LLC-1:
$$R = -(CH_2)_{11}O$$

OR LLC-2: $R = -(CH_2)_{10}$

OR LLC-3: $R = -(CH_2)_{11}$





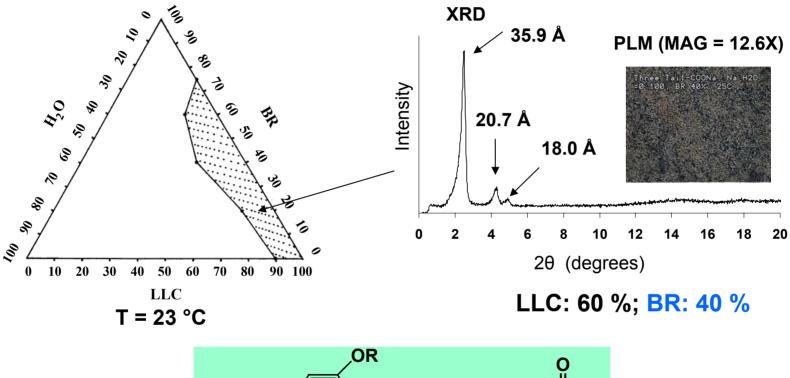
• All three LLCs form H_{II} phase.

Smith, R. C.; Fischer, W. M.; Gin, D. L. J. Am. Chem. Soc. 1997, 119, 4092. Hoag, B. P.; Gin, D. L. Macromolecules 2000, 33, 8549.

Blending Procedure and Structure Determination

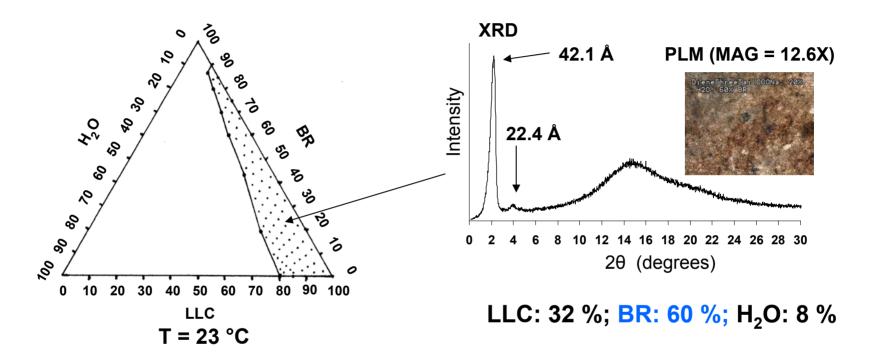
- LLCs and H₂O were mixed and centrifuged three times (3800 RPM, 15 min.).
- Add the LLCs obtained in the above step with BR precursor solution (15 wt % in hexane) and then mix/centrifuge three times (3800 RPM, 15 min.).
- Phase structures of samples were determined by X-ray diffraction (XRD) and polarized light microscopy (PLM).

Phase Diagram of LLC-1 / BR System



- LLC-1 retains H_{II} phase even with BR content as high as 75 wt %.
- Retention of H_{II} structure upon photo-initiated radical polymerization.

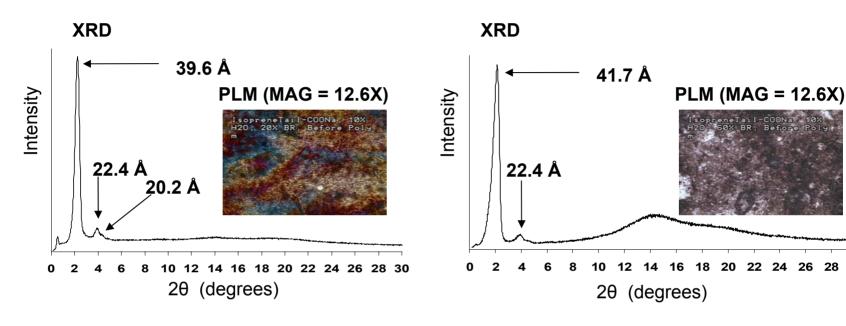
Phase Diagram of LLC-2 / BR system



$†$
Na † OOC \longrightarrow OR $R = -(CH_2)_{10}$ OR LLC - 2

- LLC-2 retains H_{II} phase even with BR content as high as 86 wt %.
- Retention of H_{II} structure upon photo-cross-linking.

XRD Profiles of LLC-3 / BR System



LLC: 72 %; BR: 20 %; H₂O: 8 %

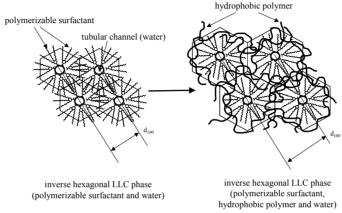
$†$
Na $^{-}$ OOC \longrightarrow OR $R = -(CH_2)_{11}$ OR LLC - 3

LLC: 36 %; BR: 60 %; H₂O: 4 %

- LLC-3 retains H_{II} phase after mixing with up to 70 wt % BR.
- Retention of H_{II} structure upon photopolymerization.

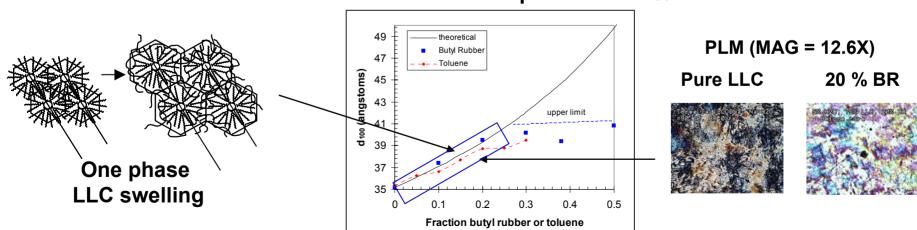
LLC / BR Microstructure Control: Swelling

Swelling of the LLC phase



Experimental vs. theoretical swelling:

Both toluene and butyl rubber can swell the theoretical maximum up to about 25%

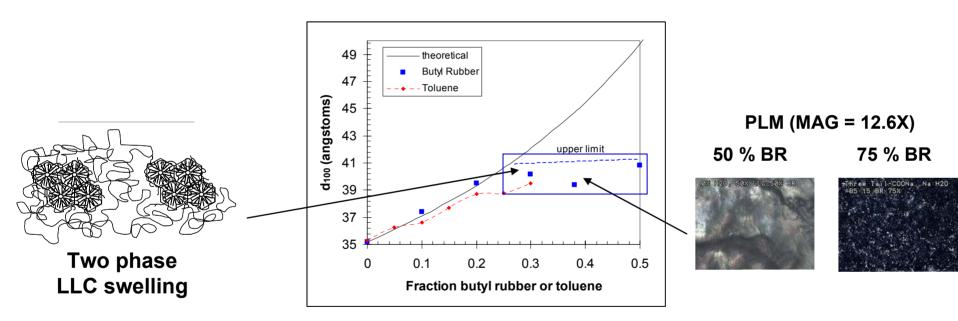


Unit cell of LLC phase increases with increasing BR content, up to 25 wt %.

LLC / BR Microstructure Control: Phase Separation

Experimental vs. theoretical swelling:

Both toluene and butyl rubber can swell the theoretical maximum up to about 25%

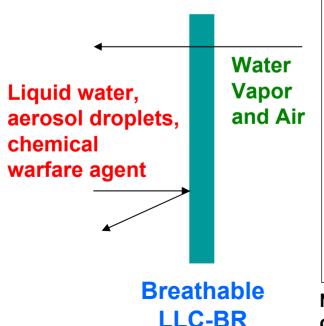


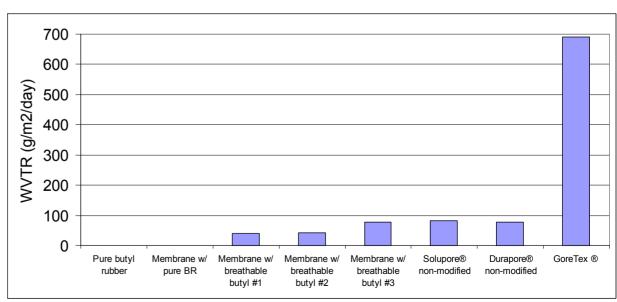
• Above 25 wt % BR, BR chains go between LLC domains, and the phase separation occurs.

Water Permeation Through LLC-BR Nanocomposites

Breathable BR membranes were produced using an ultraporous membrane support. The ultimate goal is to allow water vapor transport, while preventing chemical warfare agent (CWA) penetration.

- Water vapor transmission rate (WVTR) through membranes.
- 25 °C, 80% relative humidity to 0%.
- Mass transfer limited by ultraporous membrane support.
- Breathable BR composite membranes have 10% of the WVTR ("breathability") of GoreTex®

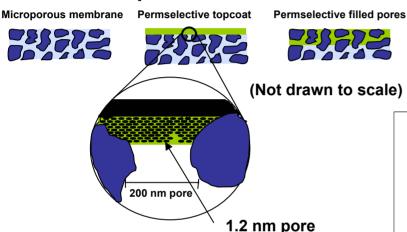




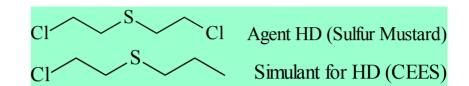
Note: GoreTex® has a higher reported WVTR for other testing conditions. The results above are for comparison purposes only.

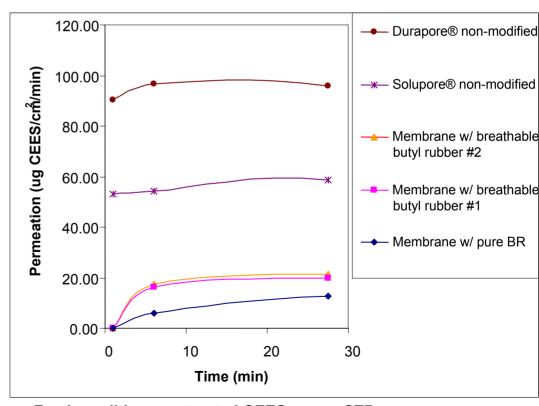
Chemical Warfare Agent Simulant Permeation Through Breathable LLC - BR Nanocomposites

Composite membranes



- Composite membranes were prepared with LLC / BR composites with varying amounts of BR.
- CEES permeation is compared for un-modified membrane supports and a composite membrane with either a breathable BR or a pure BR top-coat.
- Note that if the membranes were defect free, the pure BR should exhibit a higher resistance to CEES permeation than observed.
- Therefore, we expect even lower permeation for LLC / BR composites (breathable BR) with improved membrane production techniques.





Feed conditions: saturated CEES vapor, STP Sweep conditions: N_2 and water vapor (25% rel. hum. At 25°C)

Summary and Future Work

Summary

- Nanoporous polymer composites were prepared by blending and copolymerizing LLC monomers with commercial BR polymer.
- The LLCs in the resulting polymer composites form the H_{II} phase and allow air and water vapor to permeate, but retard permeation of chemical agent simulants.

Future Work

 More detailed investigations on the vapor permeation of the LLC-BR composites are in progress.

Acknowledgments

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